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4-(Methylamino)pyridine

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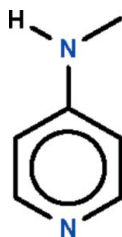
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Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.044; wR factor = 0.102; data-to-parameter ratio = 9.1.

The non-H atoms of the title compound, $\text{C}_6\text{H}_8\text{N}_2$, lie in a common plane (r.m.s. deviation = 0.034 Å). In the crystal, adjacent molecules are linked by intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds into a zigzag chain running along the c axis.

Related literature

For the non-linear optical activity of co-crystals with substituted 4-nitrophenol, see; Huang *et al.* (1997). For the crystal structure of 4-aminopyridine, see: Anderson *et al.* (2005) and for that of 4-dimethylpyridine, see: Ohms & Guth (1984).



Experimental

Crystal data

 $\text{C}_6\text{H}_8\text{N}_2$ $M_r = 108.14$ Orthorhombic, $Pna2_1$ $a = 6.5645$ (18) Å $b = 7.1230$ (19) Å $c = 12.489$ (4) Å $V = 584.0$ (3) Å³ $Z = 4$ Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹ $T = 100$ K
 $0.12 \times 0.12 \times 0.02$ mm

Data collection

Bruker SMART APEX
diffractometer
5127 measured reflections707 independent reflections
521 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.093$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.044$
 $wR(F^2) = 0.102$
 $S = 0.99$
707 reflections
78 parameters
2 restraintsH atoms treated by a mixture of
independent and constrained
refinement
 $\Delta\rho_{\text{max}} = 0.18$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.20$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|--|--------------|--------------------|-------------|----------------------|
| $\text{N1}-\text{H1}\cdots\text{N2}^i$ | 0.88 (1) | 2.06 (1) | 2.930 (3) | 168 (3) |

Symmetry code: (i) $-x, -y, z - \frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

I thank the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5213).

References

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 Ohms, U. & Guth, H. (1984). *Z. Kristallogr.* **166**, 213–219.
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supplementary materials

Acta Cryst. (2010). E66, o823 [doi:10.1107/S1600536810008986]

4-(Methylamino)pyridine

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Comment

The amino nitrogen atom in 4-aminopyridine, a drug used for treating multiple sclerosis, is pyramidal; the amino group engages in a N–H \cdots N hydrogen bonding interaction with adjacent pyridyl rings to generate a chain. The amino group uses its other nitrogen atom to form an N–H $\cdots\pi$ interaction with other pyridyl rings (Anderson *et al.*, 2005).

In the title monomethyl-substituted analogue (Scheme I, Fig. 1), all non-hydrogen atoms lie in a common plane. However, the amino nitrogen atom is slightly pyramidal, this being displaced out of the trigonal plane by 0.18 (2) Å (Σ angles 353 °). On the other hand, the amino nitrogen atom in 4-dimethylaminopyridine has unambiguously planar configuration (Ohms & Guth, 1984). In the present structure, adjacent molecules are linked by an N–H \cdots N hydrogen bond to generate a helical chain motif (Table 1).

The compound belongs to a non-centrosymmetric space group, a feature that may render it useful for second-harmonic generation, particularly as it co-crystal with 2-methoxy-4-nitrophenol shows NLO activity (Huang *et al.*, 1997).

Experimental

4-Methylaminopyridine, as purchased from the Aldrich Chemical Company, is a crystalline material.

Refinement

Due to the absence of anomalous scatterers, 644 Friedel pairs were merged. Carbon-bound H-atoms were placed in calculated positions (C–H 0.95 Å) and were included in the refinement in the riding model approximation, with $U(\text{H})$ set to $1.2U_{\text{eq}}(\text{C})$. The amino H-atom was located in a difference Fourier map and it was refined with a distance restraint of N–H 0.88±0.01 Å.

Figures

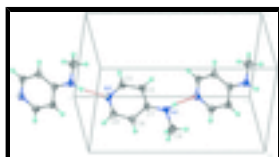


Fig. 1. Anisotropic displacement ellipsoid plot (Barbour, 2001) of a portion of the hydrogen-bonded chain structure of 4-methylaminopyridine at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

4-(Methylamino)pyridine

Crystal data

$\text{C}_6\text{H}_8\text{N}_2$

$M_r = 108.14$

$F(000) = 232$

$D_x = 1.230 \text{ Mg m}^{-3}$

supplementary materials

Orthorhombic, $Pna2_1$
Hall symbol: P 2c -2n
 $a = 6.5645$ (18) Å
 $b = 7.1230$ (19) Å
 $c = 12.489$ (4) Å
 $V = 584.0$ (3) Å³
 $Z = 4$

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 379 reflections
 $\theta = 3.3$ – 21.5°
 $\mu = 0.08$ mm⁻¹
 $T = 100$ K
Plate, colorless
 $0.12 \times 0.12 \times 0.02$ mm

Data collection

Bruker SMART APEX
diffractometer

Radiation source: fine-focus sealed tube
graphite
 ω scans
5127 measured reflections
707 independent reflections

521 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.093$
 $\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 3.3^\circ$
 $h = -7 \rightarrow 8$
 $k = -9 \rightarrow 8$
 $l = -16 \rightarrow 16$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.044$

$wR(F^2) = 0.102$

$S = 0.99$

707 reflections

78 parameters

2 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0518P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.001$

$\Delta\rho_{\text{max}} = 0.18$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.20$ e Å⁻³

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|-------------|-------------|--------------|----------------------------------|
| N1 | 0.2579 (4) | 0.0086 (3) | 0.49959 (18) | 0.0219 (6) |
| H1 | 0.177 (4) | 0.027 (5) | 0.4440 (19) | 0.036 (11)* |
| N2 | -0.0105 (4) | -0.0122 (3) | 0.8044 (2) | 0.0232 (6) |
| C1 | 0.1745 (4) | -0.0004 (4) | 0.5993 (2) | 0.0193 (6) |
| C2 | -0.0127 (4) | 0.0889 (4) | 0.6198 (2) | 0.0217 (7) |
| H2 | -0.0813 | 0.1550 | 0.5645 | 0.026* |
| C3 | -0.0957 (5) | 0.0794 (4) | 0.7215 (2) | 0.0256 (7) |
| H3 | -0.2215 | 0.1417 | 0.7336 | 0.031* |
| C4 | 0.1690 (4) | -0.0944 (4) | 0.7840 (2) | 0.0220 (7) |
| H4 | 0.2337 | -0.1593 | 0.8411 | 0.026* |
| C5 | 0.2673 (4) | -0.0917 (4) | 0.6858 (2) | 0.0207 (7) |
| H5 | 0.3959 | -0.1511 | 0.6773 | 0.025* |

| | | | | |
|-----|------------|-------------|------------|------------|
| C6 | 0.4412 (4) | -0.0966 (5) | 0.4736 (2) | 0.0302 (8) |
| H6A | 0.5534 | -0.0546 | 0.5195 | 0.045* |
| H6B | 0.4772 | -0.0753 | 0.3984 | 0.045* |
| H6C | 0.4168 | -0.2307 | 0.4853 | 0.045* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|----|-------------|-------------|-------------|--------------|--------------|--------------|
| N1 | 0.0237 (13) | 0.0242 (14) | 0.0178 (12) | 0.0021 (12) | -0.0023 (10) | 0.0025 (12) |
| N2 | 0.0277 (14) | 0.0232 (12) | 0.0189 (11) | 0.0006 (12) | 0.0005 (11) | -0.0001 (13) |
| C1 | 0.0189 (15) | 0.0192 (15) | 0.0198 (14) | -0.0048 (13) | -0.0055 (12) | -0.0010 (12) |
| C2 | 0.0231 (17) | 0.0218 (16) | 0.0201 (13) | 0.0003 (13) | -0.0031 (12) | 0.0033 (13) |
| C3 | 0.0293 (18) | 0.0210 (16) | 0.0264 (16) | 0.0046 (14) | 0.0013 (14) | -0.0005 (14) |
| C4 | 0.0248 (16) | 0.0223 (16) | 0.0188 (14) | -0.0016 (13) | -0.0039 (12) | 0.0009 (13) |
| C5 | 0.0196 (16) | 0.0205 (16) | 0.0219 (14) | 0.0009 (13) | -0.0024 (12) | -0.0011 (14) |
| C6 | 0.0242 (16) | 0.0408 (19) | 0.0256 (16) | 0.0051 (14) | 0.0028 (14) | 0.0005 (14) |

Geometric parameters (\AA , $^\circ$)

| | | | |
|-------------|------------|-------------|------------|
| N1—C1 | 1.362 (4) | C2—H2 | 0.9500 |
| N1—C6 | 1.454 (4) | C3—H3 | 0.9500 |
| N1—H1 | 0.88 (1) | C4—C5 | 1.386 (4) |
| N2—C4 | 1.340 (4) | C4—H4 | 0.9500 |
| N2—C3 | 1.345 (4) | C5—H5 | 0.9500 |
| C1—C5 | 1.401 (4) | C6—H6A | 0.9800 |
| C1—C2 | 1.407 (4) | C6—H6B | 0.9800 |
| C2—C3 | 1.384 (4) | C6—H6C | 0.9800 |
| C1—N1—C6 | 120.8 (2) | N2—C4—C5 | 124.8 (3) |
| C1—N1—H1 | 119 (2) | N2—C4—H4 | 117.6 |
| C6—N1—H1 | 113 (2) | C5—C4—H4 | 117.6 |
| C4—N2—C3 | 115.6 (3) | C4—C5—C1 | 119.1 (3) |
| N1—C1—C5 | 123.5 (3) | C4—C5—H5 | 120.4 |
| N1—C1—C2 | 119.8 (3) | C1—C5—H5 | 120.4 |
| C5—C1—C2 | 116.7 (3) | N1—C6—H6A | 109.5 |
| C3—C2—C1 | 119.3 (3) | N1—C6—H6B | 109.5 |
| C3—C2—H2 | 120.4 | H6A—C6—H6B | 109.5 |
| C1—C2—H2 | 120.4 | N1—C6—H6C | 109.5 |
| N2—C3—C2 | 124.5 (3) | H6A—C6—H6C | 109.5 |
| N2—C3—H3 | 117.7 | H6B—C6—H6C | 109.5 |
| C2—C3—H3 | 117.7 | | |
| C6—N1—C1—C5 | 7.2 (4) | C1—C2—C3—N2 | -0.6 (5) |
| C6—N1—C1—C2 | -174.3 (3) | C3—N2—C4—C5 | -0.5 (5) |
| N1—C1—C2—C3 | -179.8 (3) | N2—C4—C5—C1 | -1.2 (5) |
| C5—C1—C2—C3 | -1.1 (4) | N1—C1—C5—C4 | -179.4 (3) |
| C4—N2—C3—C2 | 1.5 (5) | C2—C1—C5—C4 | 2.0 (4) |

Hydrogen-bond geometry (\AA , $^\circ$)

| D—H \cdots A | D—H | H \cdots A | D \cdots A | D—H \cdots A |
|----------------|-----|--------------|--------------|----------------|
|----------------|-----|--------------|--------------|----------------|

supplementary materials

N1—H1[⋯]N2ⁱ

0.88 (1)

2.06 (1)

2.930 (3)

168 (3)

Symmetry codes: (i) $-x, -y, z-1/2$.

Fig. 1

